

A photograph of several glass Erlenmeyer flasks containing a bright yellow liquid, likely a gold solution, arranged in a laboratory setting. The flasks are on a reflective surface, and the background is slightly blurred, showing other laboratory equipment.

CHALLENGES IN 999.9 GOLD ASSAYING BY CUPELLATION

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22.08.2024



23-25 August 2024
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CHALLENGES IN 999.9 GOLD ASSAYING BY CUPELLATION

Standard	Analytical method	Au Range		Target Repeatability (2 replicates)	Repeatability on Au 999.9 ‰
		From (‰)	To (‰)		
ASTM E1335-08:2017	Fire Assay	5	998,0	Au <995‰ : 0.5 ‰ 995.0‰ ≥ Au < 999.5‰: 0.16 ‰	NA
ISO 11426:2021	Fire Assay	100	999,5	Au <995‰ : 0.5 ‰ 995.0‰ ≥ Au < 999.5‰: 0.16 ‰	NA
ISO 15093:2020	ICP-OES	999,0		10% of the total impurities	0.1 ‰
ISO 18214:2024	Spark-OES	999,0		Au ≥999.9: 0.01‰ 999 < Au < 999.9: 0.05 ‰	0.01 ‰
ISO 5724:2023	ICP-MS	999,99		25% of the total impurities	NA

NA = Out of scope

CHALLENGES IN 999.9 GOLD ASSAYING BY CUPELLATION

Normative	Method	Au ‰													
		(0‰)	5‰	100‰	500‰	900‰	950‰	998‰	998.5‰	999‰	999.5‰	999.9‰	999.99‰	(1000 ‰)	
ASTM E1335-08 :2017	Fire Assay		Blue												
ISO 11426:2021	Fire Assay			Blue					Blue			Red			
ISO 15093:2020	ICP-OES									Blue			Blue		
ISO 18214:2024	Spark-OES									Blue			Blue		
ISO 5724:2023	ICP-MS												Blue		

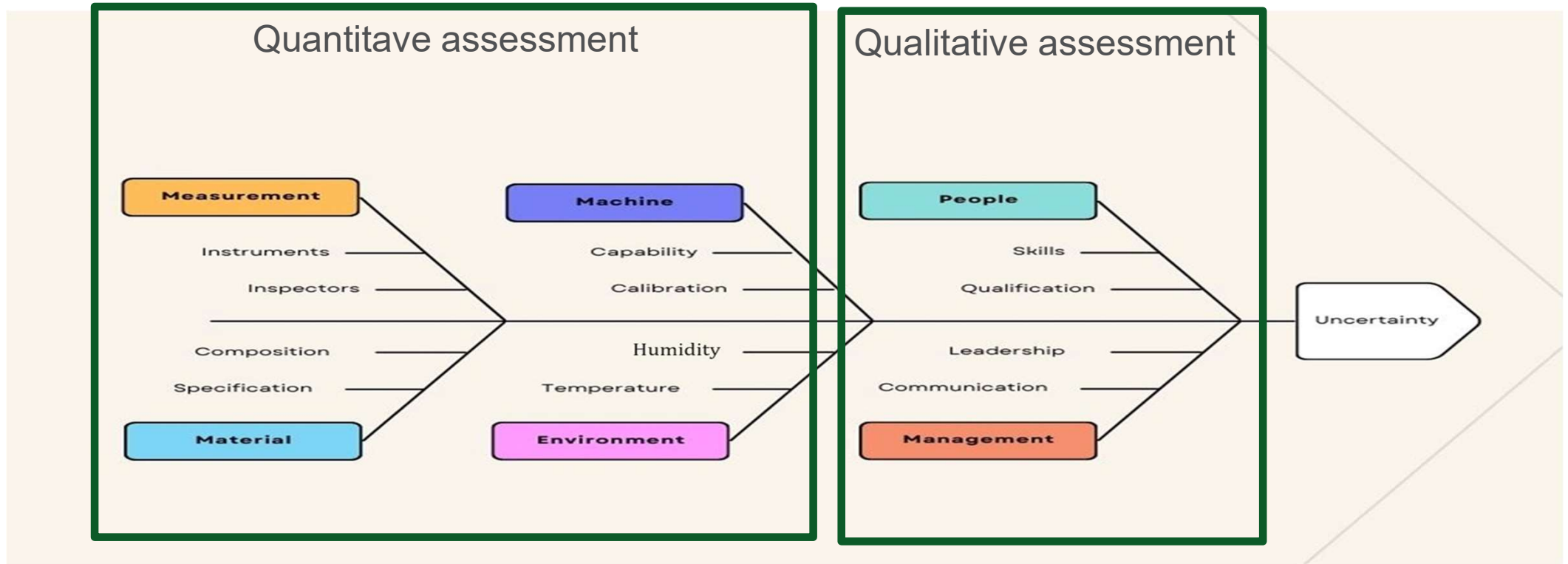
Overlapping area

In some cases the range of use is defined as «preferably»

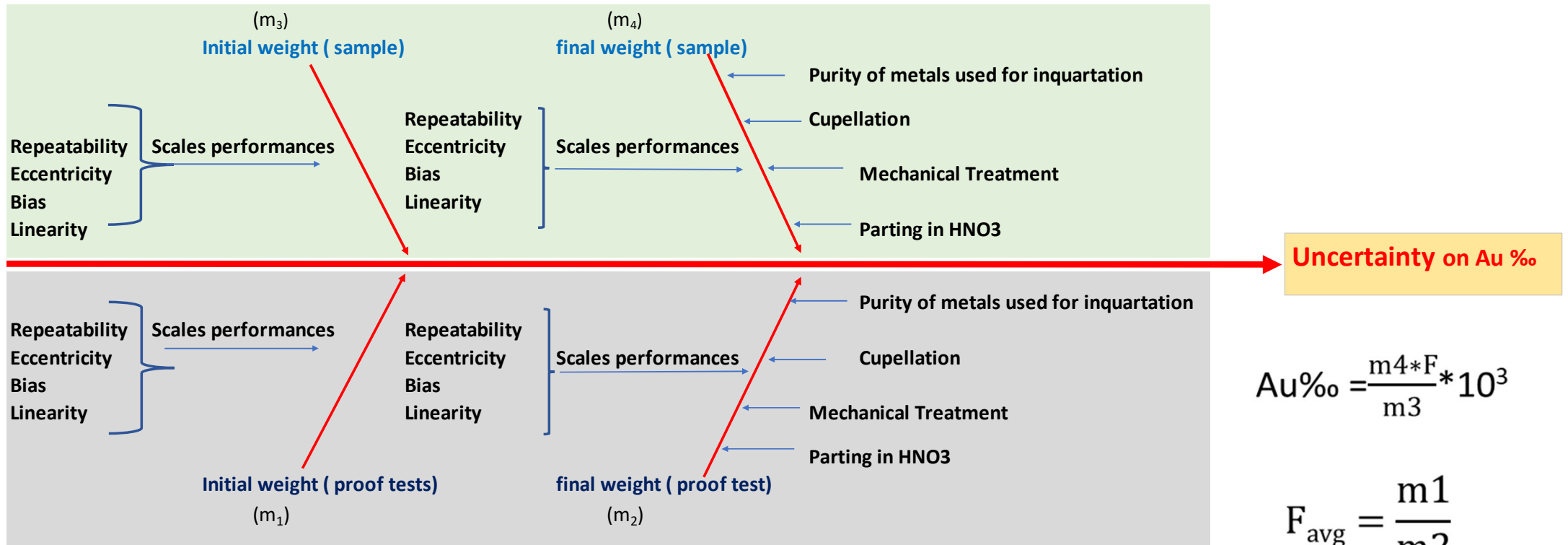
Can a small range extension in fire assay analysis be considered?

Yes but... uncertainty must be evaluated very carefully!

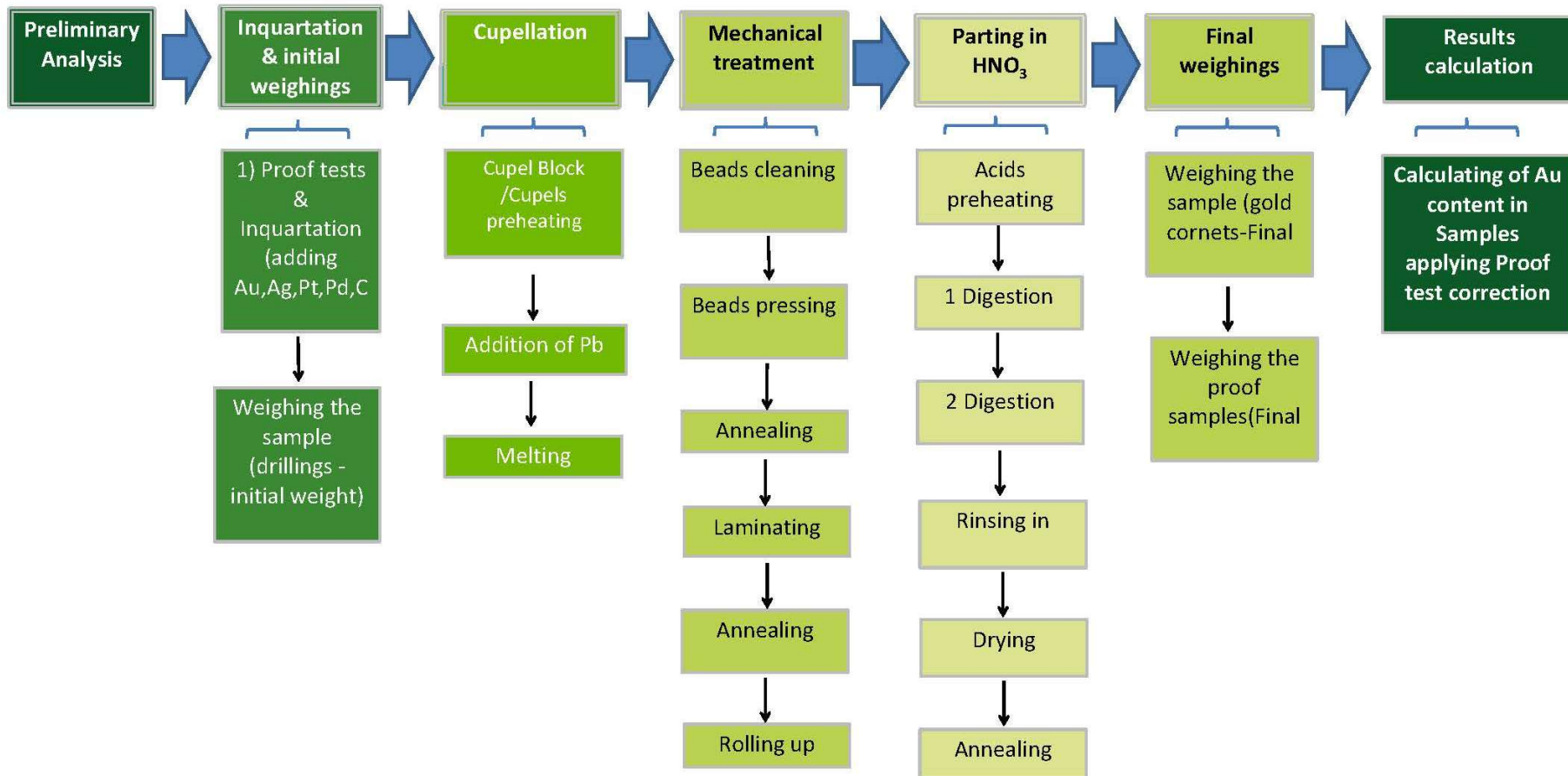
QUANTITATIVE / QUALITATIVE ASSESSMENT



CHALLENGES IN 999.9 GOLD ASSAYING BY CUPELLATION



AU CONTENT DETERMINATION VIA FIRE ASSAY - GENERAL PROCESS



INQUARTATION

Inquartation
& initial
weighings

1) Proof tests
&
Inquartation
(adding
Au,Ag,Pt,Pd,C

Weighing the
sample
(drillings -
initial weight)

Inquartation

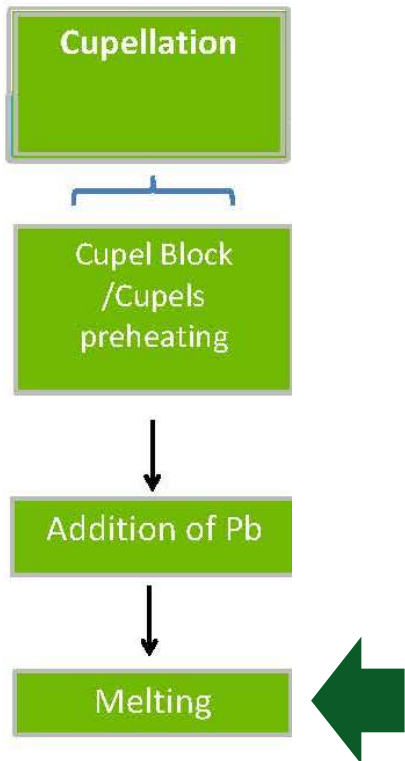
Au for proof samples min purity	999.99 ‰
Ag for proof samples min purity	999.9 ‰
Pb & Cu for proof samples min purity	999.9 ‰

Scales

Accuracy : 0.001 mg
Sample weight: 500 mg (double the standard weight)
Environment: T & RH% under control

Check the performance of the scale stated on
the certificate issued by the manufacturer!

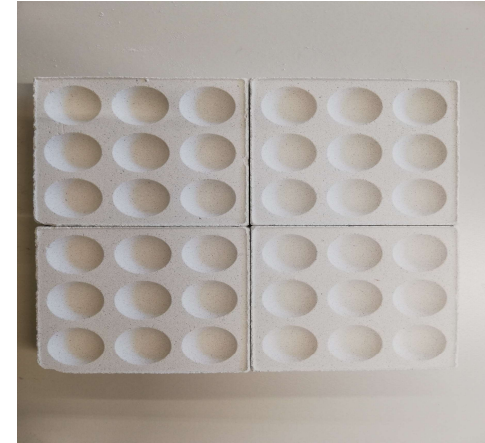
CUPELLATION



36 MgO single cupels



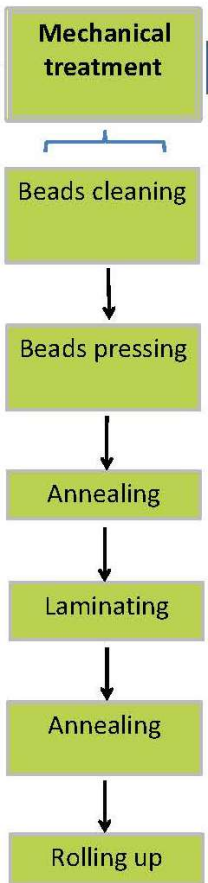
4 MgO blocks 3x3



The temperature inside the oven should be as homogeneous as possible, small variations are corrected by proof tests.

The use of an MgO blocks increases the position-related repeatability.

MECHANICAL TREATMENT, 1



Anvil & Hammer

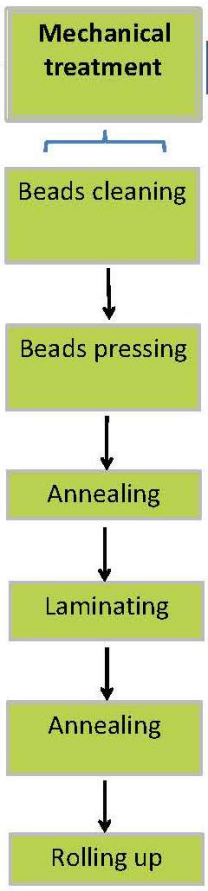


Hydraulic Press

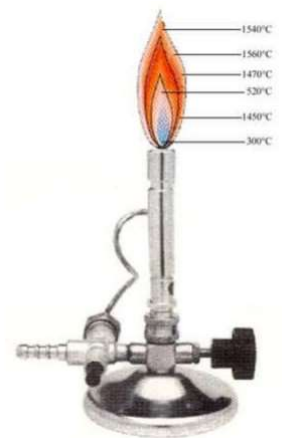


The bead resulting from cupellation must be cleaned and flattened before being rolled.
The use of anvil & hammer is historical but the surface of the hammer must be absolutely smooth and clean.
Using a press gives greater safety and repeatability

MECHANICAL TREATMENT, 2



Bunsen burner

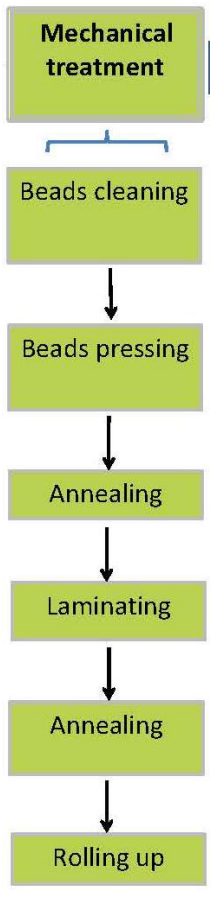


Annealing oven



The bunsen burner does not allow an effective temperature control as an annealing oven. Some areas of the flame are too hot , others too cold. In addition, samples must be handled one at a time. Samples may be damaged and have a loss of material then give results under what expected !

MECHANICAL TREATMENT, 3



Laminating

Gold beads should be prepared to a thickness of **0.12 to 0.15 mm** and rolled fairly tightly but not too tightly so that HNO₃ solutions can act effectively.

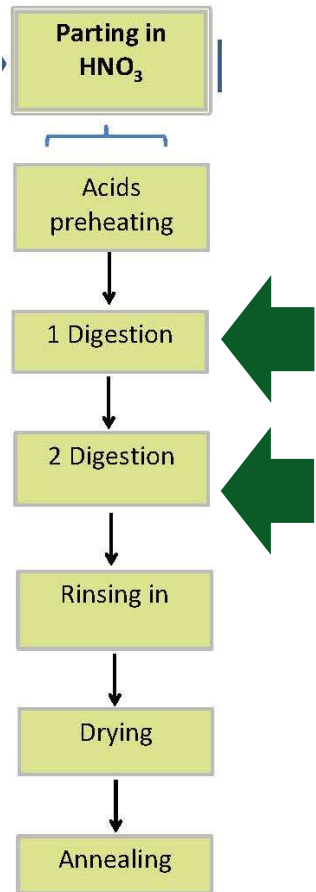
Rolls type



«S» shape «B» shape «spiral» shape

**All samples must have the same thickness !
All rolls must have the same shape !**

PARTING IN HNO_3



6 Kjeldahl flasks



6x6 Quartz basket



The use of a basket made of suitable material (quartz or platinum) instead of using traditional flasks could rationalize the HNO_3 separation operation. It allows greater control of temperature distribution during the treatment.

UNCERTAINTY

Contributes to Uncertainty		
Id	Description	Uncertainty
A	Contribute due to the Repeatability of the lab	μ_A
B	Contribute due to the Repeatability of F factor	μ_B
C	Contribute due to weighing uncertainty	μ_C
D	Contribute due to purity of Au for proof tests (999.99 ‰)	μ_D
E	Contributes due to purity of Ag for proof tests (999.9 ‰)	μ_E

Starting from U_c we can calculate the **expanded uncertainty U** taking into account a confidence level:

$$U = K * U_c$$

K=2 (95 % of confidence)

K=3 (99 % of confidence)

Accordingly to the law of propagation of the uncertainty, if the contributions are **not correlated** the variances add up with the formula:

$$U_c = \sqrt{\mu_A^2 + \mu_B^2 + \mu_C^2 + \mu_D^2 + \mu_E^2}$$

Where U_c is the Combined Standard Uncertainty

HOW TO REDUCE THE UNCERTAINTY ?

1) Reduce the uncertainty related to statistical & systematic errors :

Statistical errors : related to the standard deviations of the method
→ Improve the repeatability of the process

Systematic errors: related to an inaccuracy of the system
→ Check with Proficiency Testing and apply corrective actions

2) Increase the number of replicates

The standard uncertainty of mean value U_n of n replicates is expressed this way:

$$U_n = s / \sqrt{n}$$

Where s is the standard deviation of the method

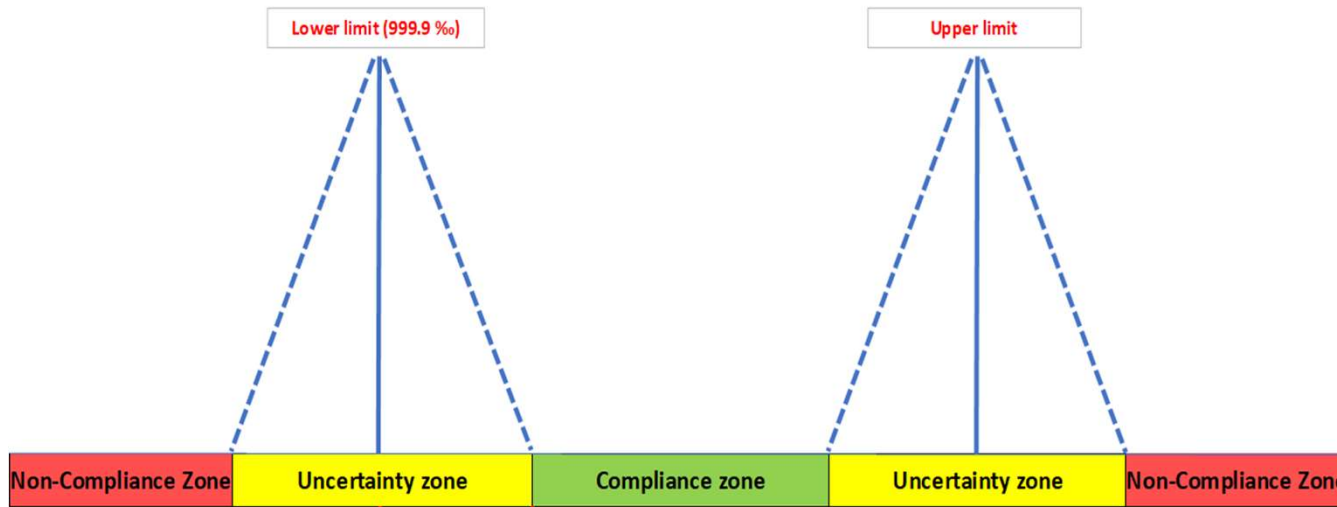
HOW MANY REPLICATES FOR SAMPLES AND PROOF TESTS?

e.g for a standard deviation of 0.07‰ (considering the same for samples and proof tests):

$$U_c = \sqrt{\mu A^2 + \mu B^2}$$

Sample replicates (A)	Proof Tests replicates (B)	Uc ‰
2	1	0.09
4	2	0.06
8	4	0.04

ISO APPROACH



$$Au_{(min)} \text{ ‰} = 999.9 \text{ ‰} + \text{Uncertainty (U = K*Uc)}$$

ISO Approach

Measurement uncertainty must be covered by the party assuming the duty of proving conformity or non-conformity and consequently carries out the measurement



Uncertainty is a cost!

CONCLUSIONS

- 1) Fire assay analysis on Au 999.9 ‰ can be considered as an alternative to spectrometric methods but only after **careful assessment of the measurement uncertainty**
- 2) An **increase in the number of standard repetitions** (e.g. ISO 11426, ASTM E1335-08) must be taken into consideration

References & Bibliography

www.iso.org International Organization for Standardization

www.astm.org American Society for Testing and Materials International

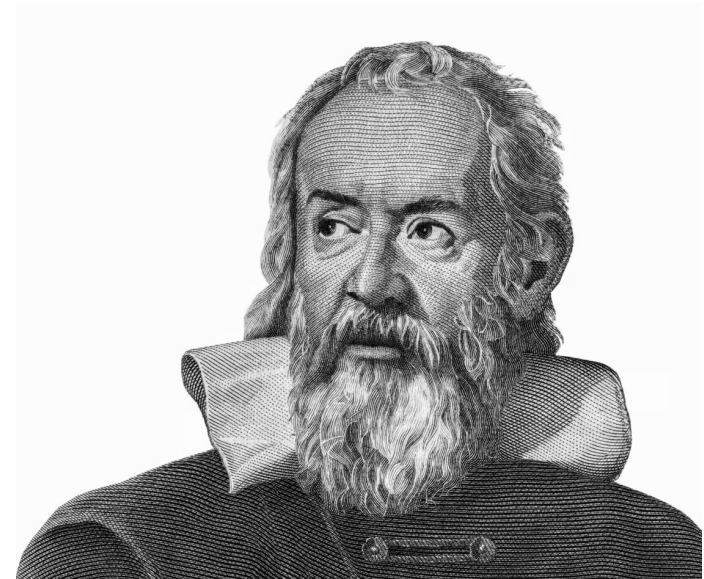
www.eurachem.org Eurachem/Citac Guide 2012

www.lbma.org.uk Assaying & Refining Conference 2013-2015 Dr A .Ruffoni

Thank you for your attention

*“Count what is countable,
measure what is measurable
and make measurable what is not”*

Galileo Galilei (1564-1642)



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